LEAD BY FIELD PORTABLE XRF

7702

MW: 207.19 (Pb) 223.19 (PbO)

Pb

CAS: 7439-92-1 (Pb) 1317-3608 (PbO) RTECS: OF7525000 (Pb) OG1750000 (PbO)

METHOD: 7702, Issue 1	EVALUATION: FULL	Issue 1: 15 January 1998
OSHA: 0.05 mg/m³ NIOSH: <0.1 mg/m³; blood lead ≤ 60 μg/100 g ACGIH: 0.05 mg/m³; BEI blood 30 μg/ 100 ml	PROPERTIES:	soft metal; d 11.3 g/mL @ 20 °C; MP 327.5°C; BP 1740°C; valences +2, +4 in salts

SYNONYMS: elemental lead and lead compounds except alkyl lead

SAMPLING		MEASUREMENT	
SAMPLER:	FILTER (0.8-µm, 37-mm, mixed cellulose ester membrane)	TECHNIQUE:	X-RAY FLUORESCENCE (XRF), PORTABLE, L-SHELL EXCITATION (e.g., ¹⁰⁹ Cd source)
FLOW RATE: 1 to 4 L/min			NOTE: Performance parameters are based upon research conducted with the
VOL-MIN: -MAX:	570 L @ 30.0 μg/m³ [1] 1900 L @ 9.0 μg/m³		NITON® 700 XRF [1].
SHIPMENT:	routine	ANALYTE:	lead
SAMPLE STABILITY:	stable	CALIBRATION:	lead thin-film standards (Micromatter Co., or equivalent); internal instrument calibration
BLANKS:	2 to 10 field blanks per set	RANGE:	17 to 1500 µg of Pb per sample [1]
ACCURACY			:6 μg of Pb per sample [1]
RANGE STUDIED: 0.1 to 1514.6 µg/m ³ (as Pb) (based upon lead mass loadings)		ESTIMATED LOD:	
BIAS:	0.069 [1]		
PRECISION (S _{rt}):	0.054 @ 10.3 to 612 µg Pb per sample		
ACCURACY:	±16.4%		

APPLICABILITY: This method was evaluated for air samples on filters only. The working range of this method is 0.017 mg/m³ to 1.5 mg/m3. This is a field portable analytical method, particularly useful for the analysis of initial exposure assessment samples, or for applications where laboratory analysis is impractical. Additionally, the method is non-destructive; samples analyzed in the field can later be analyzed in a laboratory. The method is applicable to all elemental lead forms, including lead fume, and all other aerosols containing lead.

INTERFERENCES: The presence of bromine will cause XRF readings for lead to be elevated, resulting in a positive bias error. Other interferences may exist in other XRF instruments.

OTHER METHODS: Laboratory-based methods include atomic spectrophotometric methods following hot plate acid digestion: NIOSH methods 7082 (flame atomic absorption spectrophotometry) [2], 7105 (graphite furnace atomic absorption spectrophotometry) [3], and 7300 (inductively coupled plasma atomic emission spectrophotometry) [4]. A field-portable analytical method for lead air filter samples using ultrasound/ASV has been developed, NIOSH Method 7701[5]. A field-portable screening method by spot test kit has been developed, NIOSH Method 7700 [6].

REAGENTS:

1. None

EQUIPMENT:

- 1. Sampler: Mixed cellulose ester filter, 0.8-µm pore size, 37-mm diameter, with cellulose backup pad, in a closed-faced cassette filter holder.
- 2. Personal sampling pump, 1 to 4 L/min, with flexible connecting tubing.
- 3. Field portable, L-shell X-Ray Fluorescence (XRF) instrument with a Cadmium-109 source.
- Filter sleeve: thin cardboard with 37-mm dia. cut out, and covered with a light adhesive between two pieces of acetate (Mylar[™]) (NITON, Bedford, MA, or equivalent). NOTE: Material must be transparent to X-ray.
- Filter test platform to hold the filter (specific to instrument).
- 6. Forceps
- 7. Thin film standard reference materials from 15 μ g/cm² to 150 μ g/cm² (Micromatter Co., Deer Harbor, WA), or equivalent [7,8].

SPECIAL PRECAUTIONS: None

SAMPLING:

- 1. Calibrate each sampling pump with a representative sampler in line.
- Sample at an accurately known flow rate (1 to 4 L/min) for a total sample size of approximately 1000 L. Do not exceed a filter loading of 2 mg total dust.

SAMPLE PREPARATION:

- 3. With forceps, transfer the MCE filter without the backup pad to a filter sleeve. The sleeve material must be transparent to X-rays (see EQUIPMENT, Item 4).
 - NOTE: Take special care when removing the filter from the backup pad to avoid loss of leadcontaining dust.
- 4. Place the filter into 37-mm opening and seal with Mylar[™] film to prevent losses and allow undisturbed analysis of the filter.
- 5. Place the sealed filter onto the filter test platform of the instrument for analysis.
- NOTE: The NITON® 700 Series XRF has a filter tesplatform that allows for three readings with no substrate effect.

CALIBRATION AND QUALITY CONTROL:

- 6. Start XRF and allow a 30-minute warm-up period. The instrument will conduct an internal self-calibration.
- 7. Using thin film standards [8], verify the internal calibration to withit 5% of the calibration standard. Use a minimum of three standards at concentrations of 15 μg/cm², 150 μg/cm², and one standard concentration between these two values.
- Restart the instrument as needed to assure instrument accuracy prior to sample analysis.
 NOTE: When the thin film standard measurements are not within the specified parameters, the instrument may need to be recalibrated at the factory.
 - Instrument may need to be recalibrated at the factory.
- 9. Analyze one thin film standard every 2 hours to check for instrument drift.
- 10. Repeat step 7 when all analyses are completed as a post-calibration check.

MEASUREMENT:

- 11. Set instrument parameters and analyze filter samples as specified by the manufacturer. The following measurement technique is based upon the NITON® 700 XRF.
 - a. Analyze the middle of the sample filter first (see Figure 1, M).
 - b. Allow the instrument to take a one source-minute reading (This may take longer than one real-time minute, depending upon the source strength). A one source-minute reading will assure the accurate L-shell reading necessary for the analysis of lead air filter samples.
 - c. Analyze the filter sample at the top of the filter for one source minute (see Figure 1, T).
 - d. Analyze the filter sample at the bottom of the filter for one source minute (see Figure 1, B).
 - e. The instrument software uses an algorithm that converts the three readings in μg/cmto an analytical result in μg of lead per sample. This result will be displayed following the third filter reading [1].
 - f. Analyze one standard every 2 hours (step 8).
 - g. Repeat three-reading calibration check following completion of analyses (step 8).

CALCULATIONS:

12. Using the measured lead concentration, W (μg), calculate the concentration, C (mg/ħ), of lead in the air volume sampled, V (L):

$$C = \frac{W}{V}, mg/m^3$$

NOTE: $\mu g/L = mg/m^3$

EVALUATION OF METHOD:

This method was validated on field samples [1] by collecting lead particulate samples from bridge lead abatement projects. Airborne concentrations of lead within the containment of a sand blasting bridge lead abatement project ranged from 1 to 10 mg/m Area samples were collected for periods of time ranging from 15 seconds to 2 hours. This sampling protocol yielded 61 filter samples with lead loadings ranging between 0.1 to 1514.6 µg of lead per sample. Four personal samples were collected from a hand-scraping bridge lead abatement project for a total sample size of 65. The samples were first analyzed using a non-destructive, field portable XRF method. Samples subsequently were subjected to confirmatory analysis by the laboratory based NIOSH method 7105, Lead by GFAAS [3]. The method was statistically evaluated according to the NIOSH Guidelines for Air Sampling and Analytical Method Development and Evaluation [9]. The overall precision $\hat{\mathbf{S}}_{rT}$ of the XRF method was calculated at 0.054 with a 95% confidence interval (CI) of 0.035 to 0.073, and the bias was 0.069 with a 95% CI of 0.006 to 1.515. The XRF method accuracy was determined to be ± 16%; however, at the upper 90% CI, the accuracy is $\pm 27\%$. Since the confidence interval includes the $\pm 25\%$, meeting the NIOSH accuracy criteria of ± 25% is inconclusive. However, the samples used to evaluate this method were field samples. Laboratory prepared aerosol samples would be expected to give better precision. Additionally, the XRF method is non-destructive; samples analyzed in the field can subsequently be analyzed in a laboratory using a method with greater accuracy, as needed. The filter sleeve used with the NITON® 700 Series XRF used a Mylar film to cover and seal the 37-mm filter. The lead particulate on the surface of the filter came into contact with the Mylar[™] film. Both the Mylar[™] film and the filter were digested with nitric acid and hydrogen peroxide as is specified in NIOSH Method 7105 [3].

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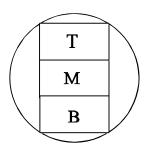


Figure 1: Analysis of a 37-mm filter (XRF windows identified as M, T,

using XRF and B are 2 cm x 1 cm)