

# Skin Notation (SK) Profile

## Chlordane

[CAS No. 57-74-9]

DRAFT

**Department of Health and Human Services**  
Centers for Disease Control and Prevention  
National Institute for Occupational Safety and Health

*This information is distributed solely for the purpose of pre dissemination peer review under applicable information quality guidelines. It has not been formally disseminated by the National Institute for Occupational Safety and Health. It does not represent and should not be construed to represent any agency determination or policy.*

## Disclaimer

Mention of any company or product does not constitute endorsement by the National Institute for Occupational Safety and Health (NIOSH). In addition, citations to Web sites external to NIOSH do not constitute NIOSH endorsement of the sponsoring organizations or their programs or products. Furthermore, NIOSH is not responsible for the content of these Web sites.

## Ordering Information

To receive this document or information about other occupational safety and health topics, contact NIOSH:

Telephone: 1-800-CDC-INFO (1-800-232-4636)

TTY: 1-888-232-6348

E-mail: [cdcinfo@cdc.gov](mailto:cdcinfo@cdc.gov)

Or visit the NIOSH Web site: [www.cdc.gov/niosh](http://www.cdc.gov/niosh)

For a monthly update on news at NIOSH, subscribe to *NIOSH eNews* by visiting [www.cdc.gov/niosh/eNews](http://www.cdc.gov/niosh/eNews).

DHHS (NIOSH) Publication No. XXX

## Foreword

As the largest organ of the body, the skin performs multiple critical functions, such as serving as the primary barrier to the external environment. For this reason, the skin is often exposed to potentially hazardous agents, including chemicals, which may contribute to the onset of a spectrum of adverse health effects ranging from localized damage (e.g., irritant contact dermatitis and corrosion) to induction of immune-mediated responses (e.g., allergic contact dermatitis and pulmonary responses), or systemic toxicity (e.g., neurotoxicity and hepatotoxicity). Understanding the hazards related to skin contact with chemicals is a critical component of modern occupational safety and health programs.

In 2009, the National Institute for Occupational Safety and Health (NIOSH) published *Current Intelligence Bulletin (CIB) 61 – A Strategy for Assigning New NIOSH Skin Notations* [NIOSH 2009-147]. This document provides the scientific rationale and framework for the assignment of multiple hazard-specific skin notations (SK) that clearly distinguish between the systemic effects, direct (localized) effects, and immune-mediated responses caused by skin contact with chemicals. The key step within assignment of the hazard-specific SK is the determination of the hazard potential of the substance, or its potential for causing adverse health effects as a result of skin exposure. This determination entails a health hazard identification process that involves use of the following:

- Scientific data on the physicochemical properties of a chemical
- Data on human exposures and health effects
- Empirical data from in vivo and in vitro laboratory testing
- Computational techniques, including predictive algorithms and mathematical models that describe a selected process (e.g., skin permeation) by means of analytical or numerical methods.

This *Skin Notation Profile* provides the SK assignments and supportive data for chlordane. In particular, this document evaluates and summarizes the literature describing the hazard potential of the substance and its assessment according to the scientific rationale and framework outlined in CIB 61. In meeting this objective, this *Skin Notation Profile* intends to inform the audience—mostly occupational health practitioners, researchers, policy- and decision-makers, employers, and workers in potentially hazardous workplaces—so that improved risk-management practices may be developed to better protect workers from the risks of skin contact with the chemicals of interest.

John Howard, M.D.  
Director  
National Institute for Occupational Safety and Health  
Centers for Disease Control and Prevention

# Contents

<b>FOREWORD</b> .....	<b>3</b>
<b>ABBREVIATIONS</b> .....	<b>5</b>
<b>GLOSSARY</b> .....	<b>7</b>
<b>ACKNOWLEDGMENTS</b> .....	<b>8</b>
<b>1.0 INTRODUCTION</b> .....	<b>10</b>
1.1 GENERAL SUBSTANCE INFORMATION: .....	10
1.2 PURPOSE.....	10
1.3 OVERVIEW OF SK ASSIGNMENT .....	11
<b>2.0 SYSTEMIC TOXICITY FROM SKIN EXPOSURE (SK: SYS)</b> .....	<b>11</b>
<b>3.0 DIRECT EFFECTS ON SKIN (SK: DIR)</b> .....	<b>14</b>
<b>4.0 IMMUNE-MEDIATED RESPONSES (SK: SEN)</b> .....	<b>15</b>
<b>5.0 SUMMARY</b> .....	<b>15</b>
<b>REFERENCES</b> .....	<b>17</b>
<b>APPENDIX: CALCULATION OF THE SI RATIO FOR CHLORDANE</b> .....	<b>20</b>
OVERVIEW .....	20
CALCULATION .....	22
APPENDIX REFERENCES .....	23

## Abbreviations

ACGIH	American Conference of Governmental Industrial Hygienists
ATSDR	Agency for Toxic Substances and Disease Registry
CIB	Current Intelligence Bulletin
cm <sup>2</sup>	square centimeter(s)
cm/hr	centimeter(s) per hour
cm/s	centimeter(s) per second
DEREK	Deductive Estimation of Risk from Existing Knowledge
DIR	skin notation indicating the potential for direct effects to the skin following contact with a chemical
EC	European Commission
GHS Chemicals	Globally Harmonized System for Labelling and Classification of Chemicals
GPMT	guinea pig maximization test
HCCPD	hexachlorocyclopentadiene
hr	hour(s)
IARC (IRR)	International Agency for Research on Cancer subnotation of SK: DIR indicating the potential for a chemical to be a skin irritant following exposure to the skin
<i>kaq</i>	coefficient in the watery epidermal layer
<i>k<sub>p</sub></i>	skin permeation coefficient
<i>k<sub>pol</sub></i>	coefficient in the protein fraction of the stratum corneum
<i>k<sub>psc</sub></i>	permeation coefficient in the lipid fraction of the stratum corneum
LD <sub>50</sub>	dose resulting in 50% mortality in the exposed population
LD <sub>Lo</sub>	dermal lethal dose
LLNA	local lymph node assay
LOAEL	lowest-observed-adverse-effect level
log <i>K<sub>ow</sub></i>	base-10 logarithm of a substance's octanol-water partition
<i>M</i>	molarity
m <sup>3</sup>	cubic meter(s)
mg	milligram(s)
mg/cm <sup>2</sup> /hr	milligram(s) per square centimeter per hour
mg/kg	milligram(s) per kilogram body weight
mg/m <sup>3</sup>	milligram(s) per cubic meter
mL	milliliter(s)
mL/kg	milliliter(s) per kilogram body weight
MW	molecular weight
NIOSH	National Institute for Occupational Safety and Health
NOAEL	no-observed-adverse-effect level
NTP	National Toxicology Program
OEL	occupational exposure limit
OSHA	Occupational Safety and Health Administration
ppm	parts per million

*This information is distributed solely for the purpose of pre dissemination peer review under applicable information quality guidelines. It has not been formally disseminated by the National Institute for Occupational Safety and Health. It does not represent and should not be construed to represent any agency determination or policy.*

REL	recommended exposure limit
RF	retention factor
SEN	skin notation indicating the potential for immune-mediated reactions following exposure of the skin
SI ratio	ratio of skin dose to inhalation dose
SK	skin notation
$S_w$	solubility
SYS	skin notation indicating the potential for systemic toxicity following exposure of the skin
USEPA	United States Environmental Protection Agency
$\mu\text{Ci}$	microcurie
$\mu\text{g}$	microgram(s)
$\mu\text{g}/\text{cm}^2$	microgram(s) per square centimeter
$\mu\text{g}/\text{m}^3$	microgram per cubic meters
$\mu\text{mol}$	micromole(s)

## Glossary

**Absorption**—The transport of a chemical from the outer surface of the skin into both the skin and systemic circulation (including penetration, permeation, and resorption).

**Acute exposure**—Contact with a chemical that occurs once or for only a short period of time.

**Cancer**—Any one of a group of diseases that occurs when cells in the body become abnormal and grow or multiply out of control.

**Contaminant**—A chemical that is (1) unintentionally present within a neat substance or mixture at a concentration less than 1.0% or (2) recognized as a potential carcinogen and present within a neat substance or mixture at a concentration less than 0.1%.

**Cutaneous (or percutaneous)**—Referring to the skin (or through the skin).

**Dermal**—Referring to the skin.

**Dermal contact**—Contact with (touching) the skin.

**Direct effects**—Localized, non-immune-mediated adverse health effects on the skin, including corrosion, primary irritation, changes in skin pigmentation, and reduction/disruption of the skin barrier integrity, occurring at or near the point of contact with chemicals.

**Immune-mediated responses**—Responses mediated by the immune system, including allergic responses.

**Sensitization**—A specific immune-mediated response that develops following exposure to a chemical, which, upon re-exposure, can lead to allergic contact dermatitis (ACD) or other immune-mediated diseases such as asthma, depending on the site and route of re-exposure.

**Substance**—A chemical.

**Systemic effects**—Systemic toxicity associated with skin absorption of chemicals after exposure of the skin.

## Acknowledgments

This document was developed by the Education and Information Division (Paul Schulte, Ph.D., Director). G. Scott Dotson, Ph.D., was the project officer for this document, assisted in great part by Naomi Hudson, Dr.P.H., Loren Tapp, M.D., and Berran Yucesoy, Ph.D. The basis for this document was a report (*Toxicology Excellence for Risk Assessment [TERA]*) contracted by NIOSH and prepared by Bernard Gadagbui, Ph.D., and Andrew Maier, Ph.D.

For their contribution to the technical content and review of this document, special acknowledgment is given to the following NIOSH personnel:

### **Denver Field Office**

Eric Esswein, M.Sc.

### **Division of Applied Research and Technology**

Clayton B'Hymer, Ph.D.

John Snawder, Ph.D.

Mark Toraason, Ph.D.

### **Division of Respiratory Disease Studies**

Gregory A. Day, Ph.D.

Aleksander Stefaniak, Ph.D.

### **Division of Surveillance, Hazard Evaluations, and Field Studies**

Matt Dahm, M.Sc.

Todd Niemeier, M.Sc.

Aaron Sussell, Ph.D.

### **Education and Information Division**

Devin Baker, M.Ed.

Charles L. Geraci, Ph.D.

Thomas J. Lentz, Ph.D.

Richard Niemeier, Ph.D.

Ralph Zumwalde, M.Sc.

### **Health Effects Laboratory Division**

Stacey Anderson, Ph.D.

H. Fredrick Frasch, Ph.D.

Vic Johnson, Ph.D.

Michael Luster, Ph.D.

Anna Shvedova, Ph.D.

Paul Siegel, Ph.D.

**National Personal Protective Technology Laboratory**

Heinz Ahlers, J.D., M.Sc.

Angie Shepherd

**Office of Surveillance, Epidemiology and Laboratory Services/Epidemiology and Analysis Program Office**

Barbara Landreth, M.A.

In addition, special appreciation is expressed to the following individuals for serving as independent, external reviewers and providing comments that contributed to the development or improvement of this document:

- G. Frank Gerberick, Ph.D., The Procter and Gamble Company, Cincinnati, Ohio
- Dori Germolec, Ph.D., National Toxicology Program, National Institute for Environmental Health Sciences, Research Triangle, North Carolina
- Ben Hayes, M.D., Ph.D., Division of Dermatology, Vanderbilt School of Medicine, Nashville, Tennessee
- Jennifer Sahmel, M.Sc., CIH, ChemRisk, Boulder, Colorado
- James Taylor, M.D., Industrial Dermatology, The Cleveland Clinic, Cleveland, Ohio

# 1.0 Introduction

## 1.1 General Substance Information

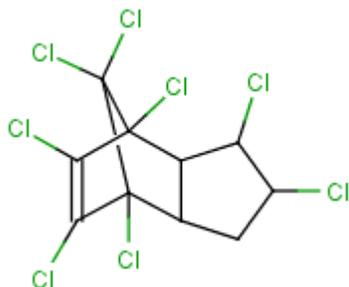
**Chemical:** Chlordane

**CAS No:** 57-74-9

**Molecular weight (MW):** 409.8

**Molecular formula:** C<sub>10</sub>H<sub>6</sub>Cl<sub>8</sub>

**Structural formula:**



**Synonyms:** Chlordan, Chlordano, 1,2,4,5,6,7,8,8-Octachloro-3a,4,7,7a-tetrahydro-4,7-methanoindane

**Uses:** Chlordane was historically used as pesticide, but all commercial uses of the substance were canceled in the United States [53 Fed. Reg. 11798 (1988)]. No information was available to determine current volume of chlordane produced in the United States.

## 1.2 Purpose

This skin notation profile presents (1) a brief summary of epidemiological and toxicological data associated with skin contact with chlordane and (2) the rationale behind the hazard-specific skin notation (SK) assignment for chlordane. The SK assignment is based on the scientific rationale and logic outlined in the *Current Intelligence Bulletin (CIB) #61: A Strategy for Assigning New NIOSH Skin Notations* [NIOSH 2009]. The summarized information and health hazard assessment are limited to an evaluation of the potential health effects of dermal exposure to chlordane. A literature search was conducted through October 2012 to identify information on chlordane, including but not limited to data relating to its toxicokinetics, acute toxicity, repeated-dose systemic toxicity, carcinogenicity, biological system/function-specific effects (including reproductive and developmental effects and immunotoxicity), irritation, and sensitization. Information was considered from studies of humans, animals, or

appropriate modeling systems that are relevant to assessing the effects of dermal exposure to chlordane.

### 1.3 Overview of SK Assignment

Chlordane is potentially capable of causing numerous adverse health effects following skin contact. A critical review of available data has resulted in the following SK assignment for chlordane: **SK: SYS**. Table 1 provides an overview of the critical effects and data used to develop the SK assignment for chlordane.

**Table 1. Summary of the SK Assignment for chlordane**

Skin Notation	Critical Effect	Available Data
SK: SYS	Hepatotoxicity	Limited human and sufficient animal data

## 2.0 Systemic Toxicity from Skin Exposure (SK: SYS)

Limited toxicokinetic data were identified that reported the degree of absorption of chlordane through the skin following dermal exposure. In rhesus monkeys administered 1 microcurie ( $\mu\text{Ci}$ )  $^{14}\text{C}$ -chlordane to 12 square centimeters ( $\text{cm}^2$ ) of abdominal skin for 24 hours (hr), percutaneous absorption, determined by the ratio of urinary excretion following topical application to that following intravenous administration, of 4.2% was reported when the substance was applied in soil medium and 6.0% when applied in acetone [Wester et al. 1992]. In an *in vitro* study using cadaver skin, 500 micromoles ( $\mu\text{m}$ )  $^{14}\text{C}$ -chlordane was applied for 24 hr. The percent of chlordane absorbed in the skin were 0.34% of the applied dose when applied in soil and 10.8% when applied in an acetone vehicle [Wester et a. 1992]. Other studies have measured detectable levels of chlordane and its metabolites in the blood of non-occupationally exposed individuals whose homes were treated with chlordane and who had detectable levels of the substance on their skin [Hirai and Tomokuni 1993a, 1993b, 1995]. The potential of chlordane to pose a skin absorption hazard was also evaluated, with use of a predictive algorithm for estimating and evaluating the health hazards of dermal exposure to substances [NIOSH 2009]. The evaluation method compares an estimated dose accumulated in the body from skin absorption and an estimated dose from respiratory absorption associated with a reference occupational exposure limit. On the basis of this algorithm, a ratio of the skin dose to the inhalation dose (SI ratio) of 0.0018 was calculated for chlordane. An SI ratio of  $\geq 0.1$  indicates that skin absorption may significantly contribute to the overall body burden of a substance [NIOSH 2009]; therefore chlordane is not considered to be absorbed through the skin following dermal exposure. Additional information on the SI ratio and the variables used in its calculation are included in the appendix.

No estimate of the human dermal lethal dose ( $\text{LD}_{50}$ ) was identified for chlordane. Dermal  $\text{LD}_{50}$  (the dose resulting in 50% mortality in the exposed animals) values reported for rats

were 840 milligrams per kilogram body weight (mg/kg) and 690 mg/kg of technical grade chlordane in xylene for males and females, respectively [Gaines 1969]. Also, the reported LD<sub>50</sub> value for female rats treated with undiluted technical grade Chlordane was 530 mg/kg [Gaines 1969]. Application of 50 mg chlordane in cottonseed oil to the skin of rats was lethal to all rats after daily application for 3 days, while identical application in ethyl alcohol caused no lethality [Ambrose et al. 1953]. The dermal LD<sub>50</sub> values reported indicate that the vehicle in which chlordane was administered influenced its toxicity. Because the reported acute dermal LD<sub>50</sub> values for rats are lower than the critical dermal LD<sub>50</sub> value of 2000 mg/kg that identifies chemical substances with the potential for acute dermal toxicity [NIOSH 2009], chlordane is considered acutely toxic following dermal exposure.

No epidemiological studies were identified that investigated the potential of chlordane to cause systemic effects following repeated or prolonged dermal exposure. Princi and Spurbeck [1951] conducted an occupational exposure study of 34 pesticide workers involved from one to three years in the manufacture and formulation of chlorinated hydrocarbons including chlordane, aldrin and dieldrin. Twenty two of the workers were directly exposed during the manufacturing process; however, the workers were exposed mainly by inhalation to total chlorinated hydrocarbon air concentrations of up to 10 milligrams per cubic meter (mg/m<sup>3</sup>) and to an unspecified amount by dermal contact [Princi and Spurbeck 1951]. The authors observed no evidence of any adverse effects on the central nervous system (CNS), the liver, the kidneys or the hematopoietic system.

Repeat-dose and subchronic dermal toxicity studies were identified in animals. In a repeated-dose study, Frings and O'Tousa [1950] administered 0.01 milliliter (mL) of 2% chlordane solution in odorless kerosene [corresponding to 7.4 milligrams per kilograms per day (mg/kg-day)] or 0.04 mL of 2% wettable powder suspension of chlordane [corresponding to 29.6 mg/kg-day] to the skin of female mice daily for 5 days per week for 20 weeks; a group of animals received no treatment and served as the control group. The incidence of seizures rose steadily in the kerosene-chlordane group, reaching 45% after 6 weeks. No increase in the incidence of seizures was observed in mice treated with the chlordane wettable powder until 9 weeks when incidence of seizures reached 50% [Frings and O'Tousa 1950]. Decreased survival was observed in animals in the kerosene-chlordane group (3 of the 16 animals died) [Frings and O'Tousa 1950]. Histological examination of the liver, the only organ evaluated, revealed liver necrosis in both chlordane treated groups [Frings and O'Tousa 1950]. The Agency for Toxic Substances and Disease Registry (ATSDR) [1994] noted that the chlordane used in the Frings and O'Tousa [1950] study was an "early" production chlordane that contained significant amounts of the reaction intermediate hexachlorocyclopentadiene (HCCPD) in contrast to the more highly purified "later" production chlordane. It is, therefore, unclear whether or not the neurological and liver effects as well as the decreased survival were due to chlordane alone or HCCPD. To assess this, the potential of HCCPD to elicit these responses was evaluated. No repeated dermal toxicity studies were identified that evaluated the potential of HCCPD to cause neurological or liver effects or decreased survival. However, subchronic gavage bioassays with HCCPD in mice and rats [Adbo et

al. 1984] indicate that HCCPD caused liver and kidney effects; however, no seizures or neurological effects at the highest oral dose of 150 mg/kg-day in rats or 300 mg/kg-day in mice were reported. Moreover, other more recent oral toxicity studies have observed hepatic effects in animals exposed to chlordane. Among these studies, Khasawinah and Grutsch [1989a, 1989b] identified hepatic effects after feeding rats 25 ppm of chlordane for 130 weeks. The overall data suggests that chlordane itself, and not HCCPD, is likely the cause of the decreased survival, seizures, and liver effects at the dermal dose of 7.4 or 29.6 mg/kg-day observed in the Frings and O'Tousa [1950] dermal toxicity study. The lowest dose in this study (7.4 mg/kg-day) can be regarded as the LOAEL, without a NOAEL being identified. Therefore, this assessment concludes that chlordane has the potential to be systemically available and may cause hepatic effects and decreased survival, with a NOAEL lower than the critical dermal NOAEL value of 1000 mg/kg-day that identifies chemical substances with the potential for repeated-dose dermal toxicity [NIOSH 2009].

No standard toxicity or specialty studies were identified that evaluated the potential for chlordane to cause biological system/function specific effects (including reproductive and developmental effects and immunotoxicity) following dermal exposure. In a 90 day subchronic skin painting study, Datta et al. [1977] exposed male guinea pigs to 0 or 168 mg/kg-day chlordane in 1 mL of acetone on shaved abdomens under unoccluded conditions. No symptoms of insecticide poisoning or significant changes in the activity of acetylcholinesterase in the blood and brain were reported; however, histopathological examination of the testes showed mild degenerative changes. Although no other dermal studies have identified the testes as a target organ, alterations in reproductive-related behavior have been reported in male rats as a consequence of an oral chlordane exposure [Cassidy et al. 1994]. Amita Rani et al. [1992] also showed the potential for reproductive consequences on tissue distribution of chlordane or its metabolites in the ovary, uterus, and adrenals in non-pregnant rats within 30 minutes after an oral dose of 120 mg/kg heptachlor. These studies indicate the potential for chlordane and its metabolites to adversely affect reproductive processes through bioconcentration in reproductive organs. Although these results are supportive of the Datta et al. [1977] study that chlordane has the potential for functional reproductive effects, no multigenerational reproductive studies were available to confirm these effects.

Studies that evaluated the potential for chlordane to be carcinogenic following dermal exposure were identified. In a population based case-control study of white male farmers by Cantor et al. [1992], the risk of non-Hodgkins lymphoma was significantly elevated (odds ratio (OR), 1.7 [95% confidence interval (CI) 1.0-1.5] with protective clothing or equipment and OR 2.2 (95% CI 1.2-4.2) without protective clothing or equipment) in 55 men who handled chlordane compared with 68 controls. However, subjects may have been simultaneously exposed to other chemicals. In a cross-sectional epidemiological investigation conducted to assess the health status of 261 people from 85 private households previously treated with chlordane for termite control, Menconi et al. [1988] reported a statistically significant increase in the incidence of unspecified skin neoplasms. The population was exposed to indoor air levels of chlordane up to greater than 5

microgram per cubic meters ( $\mu\text{g}/\text{m}^3$ ); however, the dermal contribution to the exposure was not quantified [Menconi et al 1988]. Table 2 summarizes carcinogenic designations of multiple governmental and nongovernmental organizations for chlordane.

**Table 2. Summary of the carcinogenic designations for chlordane by numerous governmental and nongovernmental organizations**

Organization	Carcinogenic designation
NIOSH [2005]	Potential occupational carcinogen
NTP [2011]	No designation
USEPA [2012]	B2: Probable human carcinogen
GHS [European Parliament 2008]	Carcinogenicity Category 2: Suspected of causing cancer
IARC [2012]	2B: Possibly carcinogenic to humans
EC [2012]*	R40: Limited evidence of a carcinogenic effect
ACGIH [2001]	A3: Confirmed animal carcinogen with unknown relevance to humans

ACGIH = American Conference of Governmental Industrial Hygienists; EC = European Commission, Joint Research, Institute for Health and Consumer Protection; GHS = Globally Harmonized System for Labelling and Classification of Chemicals; IARC = International Agency for Research on Cancer; NIOSH = National Institute for Occupational Safety and Health; NTP = National Toxicology Program; USEPA = United States Environmental Protection Agency.

\*Date accessed.

Although the predictive mathematical model (see Appendix) did not indicate that the chemical can be absorbed through the skin, limited toxicokinetic data [**Wester et al. 1992**]<sup>1</sup> were identified that indicate that chlordane has the potential to be absorbed through the skin. The acute dermal toxicity data in rats [**Gaines 1969**] and repeated-dose studies [**Frings and O'Tousa 1950**], with support from oral toxicity studies [Khasawinah and Grutsch 1989a, 1989b], provide sufficient evidence that chlordane is absorbed through the skin, is systemically available, with the potential to cause neurological and liver effects and decreased survival. Therefore, on the basis of the data for this assessment, chlordane is assigned the SK: SYS notation.

### 3.0 Direct Effects on Skin (SK: DIR)

No human or animal *in vivo* studies for corrosivity of chlordane or *in vitro* tests for corrosivity using human or animal skin models or *in vitro* tests of skin integrity using cadaver skin were identified. Datta et al. [1977] reported hyperkeratinization, cellular degeneration, such as vacuolization and a multinucleated condition in cells of the malpighian layer in the skin of guinea pigs, without any changes being observed in the dermis after exposure to a mixture of chlordane (67 mg/kg-day) and acetone through

<sup>1</sup>References in **bold** text indicate studies that serve as the basis of the SK assignments.

dermal painting for 90 days. Ambrose et al. [1953] observed no local reactions in rats topically treated with 273 mg/kg-day chlordane in ethyl alcohol for 4 days.

Despite the extensive worker experience, no controlled studies were identified that evaluated direct skin effects of chlordane in humans. Few cases of skin irritation observed during occupational exposure involved a mixture of chlordane and oils, but not chlordane alone. No severe skin reactions were reported in the available repeat-dose dermal studies, but some effects were noted by Datta et al. [1977]. No standard skin irritation tests were identified for chlordane. There is no compelling evidence that chlordane is a potent skin irritant, however the available data in animals are insufficient to adequately evaluate the potential for chlordane to cause direct skin effects following repeated dermal exposures. Therefore, on the basis of the data for this assessment, chlordane is not assigned the SK: DIR (IRR) notation.

#### **4.0 Immune-mediated Responses (SK: SEN)**

No human patch tests or predictive tests in animals (for example, guinea pig maximization tests, murine local lymph node assays, mouse ear swelling tests) or any other studies that evaluated the potential of the substance to cause skin sensitization were identified. In a 90-day study in which guinea pigs were topically treated with chlordane in acetone, Datta et al. [1977] reported that there was no evidence of sensitization. However, it is not clear whether the protocol used could have effectively detected a sensitization response.

Limited data identified did not specifically indicate that chlordane has the potential to cause skin sensitization in humans. In addition, lack of predictive tests in animals precludes adequate evaluation of the skin sensitization potential of chlordane. Therefore, on the basis of the data for this assessment, chlordane is not assigned the SK: SEN notation.

#### **5.0 Summary**

Although the predictive mathematical model (see Appendix) did not indicate that the chemical can be absorbed through the skin, limited toxicokinetic data [**Wester et al. 1992**] were identified that provide evidence of the potential of chlordane to be absorbed through the skin following dermal exposure. Sufficient evidence was identified from acute dermal toxicity data in rats [**Gaines 1969**] and repeated-dose studies [**Frings and O'Tousa 1950**], with support from oral toxicity studies [Khasawinah and Grutsch 1989a, 1989b] indicate that chlordane is absorbed through the skin, is systemically available and toxic, with the potential to cause neurological and liver effects, and decreased survival. Insufficient data were available to adequately evaluate the potential for chlordane to

cause direct skin effects or skin sensitization. Therefore, on the basis of these assessments, chlordane is assigned a composite skin notation of **SK: SYS**.

Table 3 summarizes the skin hazard designations for chlordane previously issued by NIOSH and other organizations. The equivalent dermal designations for chlordane, according to the Global Harmonization System (GHS) of Classification and Labelling of Chemicals, is Acute Toxicity Category 4 (Hazard statement: Harmful in contact with the skin) [European Parliament 2008].

**Table 3. Summary of previous skin hazard designations for chlordane**

<b>Organization</b>	<b>Skin hazard designation</b>
NIOSH [2005]	[skin]: Potential for dermal absorption; prevent skin contact
OSHA [2012] <sup>*</sup>	[skin]: Potential for dermal absorption
ACGIH [2001]	[skin]: Based on the significant toxicity, including death, from dermal exposure of animals and humans
EC [2012] <sup>*</sup>	R21: Harmful if in contact with skin

ACGIH = American Conference of Governmental Industrial Hygienists; EC = European Commission, Joint Research, Institute for Health and Consumer Protection; NIOSH = National Institute for Occupational Safety and Health; OSHA = Occupational Safety and Health Administration.

<sup>\*</sup>Date accessed.

## References

**Note:** Asterisks (\*) denote sources cited in text; daggers (†) denote additional resources.

\*53 Fed. Reg. 11798 [1988]. Environmental Protection Agency: chlordane/heptachlor termiticides; notice of cancellation and existing stocks determination and notice of intent to suspend.

\*ACGIH [2001]. Chlordane. In: Documentation of threshold limit values and biological exposure indices 7th ed., Vol. 1. Cincinnati, OH: American Conference of Governmental Industrial Hygienists.

\*Abdo, KM; Montgomery, CA; Kluwe, WM, Farnell DR, Prejean JD [1984]. Toxicity of hexachlorocyclopentadiene: Subchronic (13-week) administration by gavage to F344 rats and B6C3F1 mice. *J Appl Toxicol* 4:75-81.

\*Ambrose AM, Christenson HC, Robbins DJ, Rather J [1953]. Toxicological and Pharmacological studies on chlordane. *Archives of Industrial Hygiene and Occupational Medicine* 7:197-210.

\*Amita Rani BE, Karanth NG, Krishnakumari MK [1992]. Accumulation and embryotoxicity of the insecticide heptachlor in the albino rat. *J Environ Biol* 13(2): 95-100.

\*ATSDR (Agency for Toxic Substances and Disease Registry) [1994]. Toxicological profile for chlordane. Atlanta, GA: U.S. Department of Health and Human Services, Public Health Service [Available at: <http://www.atsdr.cdc.gov/toxprofiles/tp.asp?id=355&tid=62>]. Accessed 06-14-13.

†Balistreri WF, Partin JC, Schubert WK [1973]. Hepatic necrosis following accidental chlordane exposure [Abstract]. *Pediatr Res* 7:319.

\*Cantor KP, Blair A, Everett G, Gibson R, Burmeister LF, Brown LM, Schuman L, Dick FR [1992]. Pesticides and other agricultural risk factors for non-Hodgkin's lymphoma among men in Iowa and Minnesota. *Cancer Res* 52:2447-2455.

\*Cassidy, R.A., C.V. Vorhees, D.J. Minnema, and L. Hastings. 1994. The effects of chlordane exposure during pre- and postnatal periods at environmentally relevant levels on sex steroid-mediated behaviors and functions in the rat. *Toxicol. Appl. Pharmacol.* 126(2): 326-337.

\*Datta KK, Gupta PC, Dikshith TSS [1977]. Effect of chlordane on the skin of male guinea pigs. In: Zaidi SH (Ed.), Environmental pollution and health: proceedings of the Internal Symposium on Industrial Toxicology, November 4-7, 1975. Lucknow, India: Industrial Toxicology Research Centre pp. 608-611.

†Derbes VJ, Dent JH, Forrest WW, Johnson MF [1955]. Fatal Chlordane Poisoning. JAMA 158:1367–1369.

\*EC (European Commission) [ND]. Chlordane. In: EINICS (European Inventory of Existing Commercial Chemical Substances) [<http://esis.jrc.ec.europa.eu/>]. Accessed 11-16-12.

\*European Parliament, Council of the European Union [2008]. Regulation (EC) No 1272/2008 of the European Parliament and of the Council of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/EEC and 1999/45/EC, and amending Regulation (EC) No 1907/2006. OJEU, Off J Eur Union L353:1–1355 [<http://eur-lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2008:353:0001:1355:EN:PDF>]. Accessed 11-16-12.

\*Frings H, O'Tousa JE [1950]. Toxicity to mice of chlordane vapor and solutions administered cutaneously. Science 111:658-660.

\*Gaines TB [1969]. Acute Toxicity of Pesticides. Toxicol and Appl Pharmacol 14:515-534.

\*Hirai Y, Tomokuni K [1993a]. Levels of Chlordane, Oxychlordane, and Nonachlor on Human Skin and in Human Blood. Bull Environ Contam Toxicol 50:316-324.

\*Hirai Y, Tomokuni K [1993b]. Relationship Between Termiticide Treatment and Human Pollution by Chlordane, Oxychlordane, and Nonachlor.

\*Hirai Y, Tomokuni K [1995]. Human Pollution by Chlordane and Physical Condition of Subjects. Bull Environ Contam Toxicol 55:840-844.

\*IARC (International Agency for Research on Cancer) [2012]. Agents reviewed by the IARC monographs. In: IARC monographs on the evaluation of carcinogenic risks to humans [<http://monographs.iarc.fr/ENG/Monographs/PDFs/index.php>]. Date accessed: 06-10-13.

\*Jacobs JJ [1949]. Chlordane and dermatitis. Pest Control 17(1): 13.

\*Khasawinah AM, Grutsch JF [1989a]. Chlordane thirty-month tumorigenicity and chronic toxicity test in rats. Regul Toxicol Pharmacol 10:95-109.

\*Khasawinah AM, Grutsch JF [1989b]. Chlordane: 24-month tumorigenicity and chronic toxicity test in mice. *Regul Toxicol Pharmacol* 10:244-254.

\*Menconi S, Clark JM, Langenberg P, Hryhorczuk D [1988]. A Preliminary Study of Potential Human Health Effects in Private Residences Following Chlordane Applications for Termite Control. *Archives of Environmental Health* 43(5):349-352.

†NIOSH [1977]. Agricultural chemicals and pesticides: a subfile of the registry of toxic effects of chemical substances 1977. By Fairchild EJ, Lewis RJ Sr. Cincinnati: U.S. Department of Health, Education, and Welfare, Center for Disease Control, National Institute for Occupational Safety and Health, DHEW (NIOSH) Publication No. 77-180.

\*NIOSH [2005]. Chlordane. In: NIOSH pocket guide to chemical hazards. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005-149 [<http://www.cdc.gov/niosh/npg/>]. Accessed 06-14-13.

\*NIOSH [2009]. Current intelligence bulletin 61: a strategy for assigning new NIOSH skin notations. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2009-147 [<http://www.cdc.gov/niosh/docs/2009-147/pdfs/2009-147.pdf>]. Accessed 06-14-13.

\*NTP [2011]. Report on Carcinogens. Twelfth Edition; U.S. Department of Health and Human Services, Public Health Service. National Toxicology Program [<http://ntp.niehs.nih.gov/ntp/roc/twelfth/roc12.pdf>]. Accessed 11-01-12.

\*OSHA [ND]. Chlordane. In: OSHA Occupational Chemical Database [<http://www.osha.gov/chemicaldata/chemResult.html?recNo=422>]. Accessed 11-16-12.

\*Princi F, Spurbeck GH [1951]. A study of workers exposed to the insecticides chlordane, aldrin, dieldrin. *Archives of Industrial Hygiene and Occupational Medicine* 3:64-72.

†Schop RN, Hardy MH, Goldberg, MT [1990]. Comparison of the activity of topically applied pesticides and the herbicide 2,4-D in two short-term in vivo assays of genotoxicity in the mouse. *Fundam Appl Toxicol* 15:666-675.

\*USEPA [2012]. Chlordane. In: Integrated risk information system [<http://www.epa.gov/iris/subst/0142.htm#noncar>]. Accessed 11-16-12.

\*Wester RC, Maibach HI, Sedlik L, Melendres J, Liao CL, DiZio S [1992]. Percutaneous absorption of [<sup>14</sup>C]chlordane from soil. *J Toxicol Environ Health* 35:269-277.

## Appendix: Calculation of the SI Ratio for Chlordane

This appendix presents an overview of the SI ratio and a summary of the calculation of the SI ratio for chlordane. Although the SI ratio is considered in the determination of a substance's hazard potential following skin contact, it is intended only to serve as supportive data during the assignment of the NIOSH SK. An in-depth discussion on the rationale and calculation of the SI ratio can be found in Appendix B of the *Current Intelligence Bulletin (CIB) #61: A Strategy for Assigning New NIOSH Skin Notations* [NIOSH 2009].

### Overview

The SI ratio is a predictive algorithm for estimating and evaluating the health hazards of skin exposure to substances. The algorithm is designed to evaluate the potential for a substance to penetrate the skin and induce systemic toxicity [NIOSH 2009]. The goals for incorporating this algorithm into the proposed strategy for assigning SYS notation are as follows:

- (1) Provide an alternative method to evaluate substances for which no clinical reports or animal toxicity studies exist or for which empirical data are insufficient to determine systemic effects.
- (2) Use the algorithm evaluation results to determine whether a substance poses a skin absorption hazard and should be labeled with the SYS notation.

The algorithm evaluation includes three steps:

- (1) determining a skin permeation coefficient ( $k_p$ ) for the substance of interest,
- (2) estimating substance uptake by the skin and respiratory absorption routes, and
- (3) evaluating whether the substance poses a skin exposure hazard.

The algorithm is flexible in the data requirement and can operate entirely on the basis of the physicochemical properties of a substance and the relevant exposure parameters. Thus, the algorithm is independent of the need for biologic data. Alternatively, it can function with both the physicochemical properties and the experimentally determined permeation coefficient when such data are available and appropriate for use.

The first step in the evaluation is to determine the  $k_p$  for the substance to describe the transdermal penetration rate of the substance [NIOSH 2009]. The  $k_p$ , which represents the overall diffusion of the substance through the stratum corneum and into the blood capillaries of the dermis, is estimated from the compound's molecular weight ( $MW$ ) and base-10 logarithm of its octanol–water partition coefficient ( $\log K_{ow}$ ). In this example,  $k_p$  is determined for a substance with use of Equation 1. A self-consistent set of units must be used, such as centimeters per hour (cm/hr), outlined in Table A1. Other model-based estimates of  $k_p$  may also be used [NIOSH 2009].

### Equation 1: Calculation of Skin Permeation Coefficient ( $k_p$ )

$$k_p = \frac{1}{\frac{1}{k_{psc} + k_{pol}} + \frac{1}{k_{aq}}}$$

where  $k_{psc}$  is the permeation coefficient in the lipid fraction of the stratum corneum,  $k_{pol}$  is the coefficient in the protein fraction of the stratum corneum, and  $k_{aq}$  is the coefficient in the watery epidermal layer. These components are individually estimated by

$$\begin{aligned}\log k_{psc} &= -1.326 + 0.6097 \times \log K_{ow} - 0.1786 \times MW^{0.5} \\ k_{pol} &= 0.0001519 \times MW^{-0.5} \\ k_{aq} &= 2.5 \times MW^{-0.5}\end{aligned}$$

The second step is to calculate the biologic mass uptake of the substance from skin absorption (skin dose) and inhalation (inhalation dose) during the same period of exposure. The skin dose is calculated as a mathematical product of the  $k_p$ , the water solubility ( $S_w$ ) of the substance, the exposed skin surface area, and the duration of exposure. Its units are milligrams (mg). Assume that the skin exposure continues for 8 hours to unprotected skin on the palms of both hands (a surface area of 360 square centimeters [ $\text{cm}^2$ ]).

### Equation 2: Determination of Skin Dose

$$\begin{aligned}\text{Skin dose} &= k_p \times S_w \times \text{Exposed skin surface area} \times \text{Exposure time} \\ &= k_p (\text{cm/hr}) \times S_w (\text{mg/cm}^3) \times 360 \text{ cm}^2 \times 8 \text{ hr}\end{aligned}$$

The inhalation dose (in mg) is derived on the basis of the occupational exposure limit (OEL) of the substance—if the OEL is developed to prevent the occurrence of systemic effects rather than sensory/irritant effects or direct effects on the respiratory tract. Assume a continuous exposure of 8 hours, an inhalation volume of 10 cubic meters ( $\text{m}^3$ ) inhaled air in 8 hours, and a factor of 75% for retention of the airborne substance in the lungs during respiration (retention factor, or RF).

### Equation 3: Determination of Inhalation Dose

$$\begin{aligned}\text{Inhalation dose} &= \text{OEL} \times \text{Inhalation volume} \times \text{RF} \\ &= \text{OEL} (\text{mg/m}^3) \times 10 \text{ m}^3 \times 0.75\end{aligned}$$

The final step is to compare the calculated skin and inhalation doses and to present the result as a ratio of skin dose to inhalation dose (the SI ratio). This ratio quantitatively indicates (1) the significance of dermal absorption as a route of occupational exposure to the substance and (2) the contribution of dermal uptake to systemic toxicity. If a substance has an SI ratio greater than or equal to 0.1, it is considered a skin absorption hazard.

## Calculation

Table A1 summarizes the data applied in the previously described equations to determine the SI ratio for chlordane. The calculated SI ratio was 0.0018. On the basis of these results, chlordane is predicted to represent a skin absorption hazard.

**Table A1. Summary of Data used to Calculate the SI Ratio for chlordane**

Variables Used in Calculation	Units	Value
<b>Skin permeation coefficient</b>		
Permeation coefficient of stratum corneum lipid path ( $k_{psc}$ )	cm/hr	0.0652
Permeation coefficient of the protein fraction of the stratum corneum ( $k_{poi}$ )	cm/hr	$7.5036 \times 10^{-6}$
Permeation coefficient of the watery epidermal layer ( $k_{aq}$ )	cm/hr	0.1235
Molecular weight ( $MW$ ) <sup>a</sup>	amu	409.78
Base-10 logarithm of its octanol–water partition coefficient ( $\text{Log } K_{ow}$ ) <sup>a</sup>	None	6.16
Calculated skin permeation coefficient ( $k_p$ )	cm/hr	0.04268
<b>Skin dose</b>		
Water solubility ( $S_w$ ) <sup>a</sup>	mg/cm <sup>3</sup>	$5.6 \times 10^{-5}$
Calculated skin permeation coefficient ( $k_p$ )	cm/hr	0.04268
Estimated skin surface area (palms of hand)	cm <sup>2</sup>	360
Exposure time	hr	8
Calculated skin dose	mg	0.0069
<b>Inhalation Dose</b>		
Occupational exposure limit (OEL) <sup>b</sup>	mg/m <sup>3</sup>	0.5
Inhalation volume	m <sup>3</sup>	10
Retention factor (RF)	None	0.75
Inhalation dose	mg	3.75
<b>Skin dose–to–inhalation dose (SI) ratio</b>	None	0.0018

<sup>a</sup>Variables identified from SRC [2009].

<sup>b</sup>The OEL used in calculation of the SI ratio for chlordane was the NIOSH recommended exposure limit (REL) [NIOSH 2005].

## **Appendix References**

NIOSH [2005]. NIOSH pocket guide to chemical hazards. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005-149 [<http://www.cdc.gov/niosh/npg/>]. Accessed 07-07-09.

NIOSH [2009]. Current intelligence bulletin 61: a strategy for assigning new NIOSH skin notations. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2009-147 [<http://www.cdc.gov/niosh/docs/2009-147/pdfs/2009-147.pdf>]. Accessed 07-07-09.

SRC [2009]. Interactive PhysProp database demo [<http://www.srcinc.com/what-we-do/databaseforms.aspx?id=386>]. Accessed 12-02-09.