S-Benzylmercapturic acid and S-phenylmercapturic acid in urine

8326

Metabolites of toluene and benzene

 S-Benzylmerce S-Phenylmerce 	capturic acid: capturic acid:	FORMULA: $C_{12}H_{15}NO_3S$ FORMULA: $C_{11}H_{13}NO_3S$	MW: 253.3 MW: 239.3	CAS: 19542-77-9 CAS: 20640-68-0
METHOD: 83	326, Issue 1	EVALUATION: Full		lssue 1: 20 May 2014
Published limits 1. Toluene: OSH 2. Benzene: OSH Other OELs: Ber over time, NIOSH updated limits a markers for tolu	and guidelines usin A and NIOSH: None HA and NIOSH: None cause data on expos H recommends refer and guidelines conce ene and benzene [1,	g these compounds as markers: e ure limits and guidelines may change ring to the following sources for erning the use of these compounds as 2,3,4].	BIOLOGICAL INDICATOR OF: PROPERTIES: 1. Solid; 1.246 g 2. Solid; 1.28 g/c	Exposure to 1. toluene and 2. benzene /cm ³ ; mp 162-163 °C cm ³ ; mp 155 °C

SYNONYMS (not all inclusive):

- 1. S-Benzylmercapturic Acid (BMA): S-benzyl-N-acetyl-L-cysteine; (2R)-2-Acetamido-3-(phenylmethylsulfanyl)propanoic acid; Alanine, N-acetyl-3-(benzylthio)-; S-phenylmethyl-N-acetyl-L-cysteine
- 2. S-Phenylmercapturic acid (PMA): S-phenyl-N-acetyl-L-cysteine; (2R)-2-Acetamido-3-(phenylsulfanyl)propanoic acid; (2R)-2-Acetylamino-3-(phenylthio)propionic acid

	SAMPLING	MEASUREMENT			
SPECIMEN: Urine		TECHNIQUE:	HIGH-PERFORMANCE LIQUID CHROMATOGRAPHY - TANDEM MASS SPECTROMETRY (HPLC/MS/MS)		
VOLUME: At least 8 m PRESERVATIVE: None	L added. Refrigerate or freeze upon	ANALYTES:	S-Benzylmercapturic acid and S-phenylmercapturic acid		
collec	tion.	EXTRACTION:	Solid-Phase Extraction (SPE) C18		
SHIPMENT: Ship cold or receipt at th	r frozen with ice or dry ice. Freeze upon ne laboratory.	COLUMN:	C18 [dimethyloctadecylsilane solid phase type, 3.5 μm particle size] (150 mm by 3 mm)		
SAMPLE STABILITY: Stable in fro	ozen urine for periods of a month or more	PHASES:	$ A = 5/95/0.1\% (v/v/v) \text{ acetonitrile/water/acetic acid} \\ B = 75/25/0.1\% (v/v/v) \text{ acetonitrile/water/acetic acid} $		
and for seve	and for several freeze/thaw cycles [5,6].		0.3 mL/min (0.4 mL/min post run)		
CONTROLS: Urine specie level expose	mens obtained from non-exposed or low ed individuals.	GRADIENT:	Time (min) vs. Mobile Phase Composition 0 to 10 0 to 40% B 10 to 18 40 to 100% B		
	ACCURACY*		18 to 20 100% B 20 to 21 100% B Flow increased to 0.4 mL/min		
RANGE STUDIED:	See Table 2		21 to 28 100% B (0.4 mL/min flow) 28 to 30 100 to 0% B (re-equilibration, 0.3 mL/min) 30 to 37 0% B (re-equilibration)		
BIAS:	Negligible	INJECTION VOLUN	1Ε: 8 μL		
OVERALL PRECISION (Ŝ _{-T})*:	See Table 2	IONIZATION SOURCE:	Electrospray at 3500 Volts and negative scan mode, nebulizer gas at 35 psi and 10 L/min flow		
ACCURACY*:	Overall recoveries (103% overall) obtained from spiked urine samples (n=48) were 103% and 106% for S-benzylmercapturic acid and S-phenylmercapturic acid, respectively. The precision as relative standard deviation was no greater than 5.0% at	DETECTOR (MS/MS	S): Dwell time = 200 msec; Fragmentor at 80 Volts; Collision energy at 8 Volts; Collision gas: nitrogen at 0.06 L/min		
		MULTIPLE REACTION MODE:Quantification mass transitions;BMA = m/z 252 \rightarrow 123, PMA = 238 \rightarrow 109, d ₃ -BMA = 257 \rightarrow 128, d ₅ -PMA = 243 \rightarrow 114			
		TOTAL RUN TIME:	Approximately 37 minute cycle time		
		CALIBRATION:	BMA and PMA solutions with internal standards		
* The definitions of prec	ision and accuracy in this method are	QUALITY CONTRO	: At least one level of spiked urine specimen prepared from a separately weighed stock solution		
those utilized by the US F	Food and Drug Administration [7].	RANGE:	0.5 to 50 ng/mL for BMA and PMA		
, , , , , , , , , , , , , , , , , , , ,		ESTIMATED LOD:	Approximately 0.2 ng/mL for BMA and PMA; by lowest standard levels, BMA and PMA = 0.5 ng/mL		
		PRECISION (Ŝ_):	See Table 2		

APPLICABILITY: BMA and PMA are metabolites of toluene and benzene, respectively. PMA is a very specific biomarker for benzene. BMA can form from exposure to other sources, such as benzyl acetate or benzyl alcohol which can be found in personal care products [8]. Both toluene and benzene are common solvents with multiple occupational uses; furthermore exposure to toluene and benzene (less commonly) can occur from environmental and other sources. This method measures the quantity of the two target metabolite analytes in urine.

INTERFERENCES: None found or identified.

OTHER METHODS: This method is based on the research of B'Hymer [9,10]. There are numerous literature methods for these two compounds, but standardized methods from governmental agencies or consensus standards organizations are not currently available, though there is also a recent method developed at CDC/National Center for Environmental Health for these analytes and others [11].

REAGENTS:

- S-Benzylmercapturic acid (BMA, N-acetyl-S-benzyl-DLcysteine, CAS no. 19542-77-9) reference standard 0.20 mg/mL stock solution in methanol. Store at 4 °C in amber vials (or in the dark).
- 2. S-Phenylmercapturic acid (PMA, phenyl-*N*-acetyl-DLcysteine, CAS no. 20640-68-0) reference standard 0.20 mg/mL stock solution in methanol. Store at 4 °C in amber vials (or in the dark).
- 3. Deuterated S-benzyl-d₅-mercapturic acid reference standard (d₅-BMA), 0.15 mg/mL stock solution in methanol. Store at 4 °C in amber vials (or in the dark).
- Deuterated S-phenyl-d₅-mercapturic acid reference standard (d₅-PMA), 0.15 mg/mL stock solution in methanol. Store at 4 °C in amber vials (or in the dark).
- 5. Internal standard spiking solution, deuterated stock solutions diluted to approximately 30 ng/mL in water
- 6. Acetone, HPLC grade or better*
- 7. Acetonitrile, HPLC grade or better*
- 8. Acetic Acid, glacial, ACS reagent grade or better*
- 9. Methanol, HPLC grade or better*
- 10. Water, doubly deionized, minimum resistivity of 18 $$M\Omega$\mathchar`-cm$
- 11. Water, HPLC grade
- 12. Acetonitrile/water (50/50%, v/v) injector rinse solution
- 13. Chromatographic matrix adjustment solution, 49/50/1% (v/v/v) acetonitrile/water/acetic acid
- 14. Mobile phase A (5/95/0.1%, v/v/v) acetonitrile/water/ acetic acid, filtered through 0.7 μm glass microfiber filters
- 15. Mobile phase B (75/25/0.1%, v/v/v) acetonitrile/water/ acetic acid, filtered through 0.7 μm glass microfiber filters
- 16. Synthetic urine substitute, if desired
- 17. Nitrogen, UHP

* See SPECIAL PRECAUTIONS*

EQUIPMENT:

- 1. High-Performance Liquid Chromatograph (HPLC) equipped with a tandem mass spectrometric detector with data collection system
- 2. HPLC column (C18 [dimethyloctadecylsilane solid phase type], 150 mm X 3 mm, 3.5 μm particle size)
- 3. Autosampler
- 4. Analytical balance, 0.1 mg readability
- 5. Rotary vacuum concentrator with cold trap and vacuum pump or analytical evaporator with nitrogen
- 6. Automatic pipet with disposable tips in the 1000 μL volume delivery range
- 7. Solid-phase extraction (SPE) cartridges, C18, 500 mg, 3 mL
- 8. SPE vacuum manifold apparatus
- 9. Culture tubes, disposable screw-top (16 X 150 mm) with PTFE-lined caps
- 10. Tubes, polypropylene, disposable screw-top, 15 mL. Polypropylene bottles and caps may also be used for sample collection.
- 11. Metal spatula
- 12. Volumetric flasks: 10, 50, 100 and 200 mL
- 13. Autosampler vials, amber, with caps and septa
- 14. Disposable glass pipets
- 15. Sintered glass filtration apparatus and 1-Liter sidearmed Erlenmeyer flask
- 16. Microfiber filters, glass, 0.7 μm ; diameter to fit the filtration apparatus
- 17. Wet or dry ice

SPECIAL PRECAUTIONS: Standard precautions should always be used when handling bodily fluids and/or extracts of bodily fluids [12]. Handle urine specimens and urine extracts using powder-free latex or nitrile gloves. Acetic acid, acetone, acetonitrile and methanol are flammable; handle with care and use in a chemical fume hood. Handle all chemicals using the required safety precautions. Reagents with manufacturer expiration dates should be observed.

SAMPLING:

- 1. Collect at least 8 mL of urine in an appropriate polypropylene tube or bottle and cap. Refrigerate at 4 °C or freeze after collection. Collect at least two urine specimens for each worker, one before the work shift and one after.
- 2. Ship the specimen stored in either wet or dry ice in an insulated container. Freeze specimens upon arrival at the laboratory and store frozen. A reminder: commercial shippers have special labeling requirements for packages containing dry ice.

SAMPLE PREPARATION:

- NOTE: BMA and PMA are somewhat light sensitive (see Evaluation of Method section). Perform the sample preparation steps in a low light environment. Extreme measures are not required.
- 3. Thaw the urine specimen to room temperature.
- 4. Mix urine specimen thoroughly to ensure homogeneity.
- 5. Transfer 4.0 mL of urine into a 16 X 150 mm (or larger) screw-capped culture tube.
- 6. Add 0.5 mL of deionized water to aid in the dissolution of solids.
- 7. Add 0.5 mL of the 30 ng/mL deuterated BMA/PMA internal standard solution.
- 8. Solid-Phase Extraction
 - NOTE: All SPE steps are performed using the vacuum manifold apparatus. The flow rate should not exceed 1 mL/min. The SPE cartridges should not be allowed to go to dryness until the end of Step 8d. Refer to manufacturers' recommendations for use of specific SPE cartridges.
 - a. Pre-wash the C18 SPE cartridge with 2 mL of acetone.
 - b. Equilibrate the SPE cartridge with 2 mL of HPLC grade water.
 - c. Load the 5-mL urine mixture and draw the sample through the cartridge.
 - d. Wash the cartridge with 1 mL of HPLC grade water. Discard any liquid collected up to this point.
 - e. Apply (or increase) the vacuum to pull most of the water from the cartridge.
 - f. Elute the analytes with 3 mL of acetone three times, collecting all of the acetone washes into a 15-mL plastic screw-capped tube.
- 9. Evaporate the 9 mL of acetone from the extracts to dryness by means of a vacuum rotary concentrator or by using a nitrogen sweep.
- 10. Cap the plastic tubes containing the dry extract and store in a refrigerator/freezer until ready for chromatographic analysis.
- 11. Prior to chromatographic analysis, dissolve the extract in 1 mL of HPLC mobile phase A and transfer the sample into an amber HPLC autosampler vial.

CALIBRATION AND QUALITY CONTROL:

- 12. Prepare a BMA/PMA standard mixture by combining 2.0 mL of each stock solution and diluting to 100 mL with deionized water to make an approximately 4 μg/mL solution concentration for each.
- BMA/PMA standard solutions: The 4 μg/mL BMA/PMA solution is diluted in water to make 4, 8, 16, 40, 80, 320 and 400 ng/mL BMA/PMA solutions.
- 14. Accurately transfer 0.500 mL of each BMA/PMA solution from step 13 into a separate HPLC autosampler vial. [This delivers 2, 4, 8, 20, 40, 160 and 200 ng of each analyte to each autosampler vial.]
- 15. Add 0.5 mL of the deuterated internal standard spiking solution [30 ng/mL] to each vial. [This delivers 15 ng of the internal standards to each autosampler vial.]

- 16. Add 0.1 mL of the chromatographic matrix adjustment solution to each standard vial. NOTE: These standard samples are *equivalent* to 0.5 to 50 ng/mL BMA/PMA urine samples [based on the original 4.0 mL urine volume extracted and placed in each autosampler vial.]
- 17. Prepare one blank urine sample without analyte spikes; alternatively, prepare one blank using a urine substitute by following steps 14-16 using 0.5 mL of the unspiked urine or substitute in step 14.
- 18. Prepare at least one quality control (QC) standard of urine or urine substitute fortified with PMA/ BMA using separately prepared stock solutions. A 10 ng/mL equivalent spike level is suggested and more than one level can be used if desired.

MEASUREMENT:

- 19. Set the high-performance liquid chromatograph according to the manufacturer's recommendations and to the conditions listed on page 8326-1. A needle rinse with 50/50% (v/v) acetonitrile/water is required to eliminate sample carry-over by the autosampler.
- 20. Set the mass spectrometric detector to multiple reaction mode (MRM) according to the manufacturer's recommendations and the conditions listed on page 8326-1. Example conditions are summarized below:

Precursor Ion	MS1 Resolution	Product Ion	MS2 Resolution	Dwell Time (msec)	Fragmentor Voltage	Collision Energy (volt)
257	unit	128	unit	200	80	8
252	unit	123	unit	200	80	8
243	unit	114	unit	200	80	8
238	unit	109	unit	200	80	8
	Precursor lon 257 252 243 238	PrecursorMS1 Resolution257unit252unit243unit238unit	Precursor lonMS1 ResolutionProduct lon257unit128252unit123243unit114238unit109	PrecursorMS1 ResolutionProduct IonMS2 Resolution257unit128unit252unit123unit243unit114unit238unit109unit	Precursor IonMS1 ResolutionProduct IonMS2 ResolutionDwell Time (msec)257unit128unit200252unit123unit200243unit114unit200238unit109unit200	Precursor IonMS1 ResolutionProduct IonMS2

Table 1. MS/MS CONDITIONS USING NEGATIVE ELECTROSPRAY IONIZATION

21. Inject 8 μ L of each sample extract, standard, QC standard, and blank. Sample chromatograms for each compound are illustrated in Figure 1.

22. Measure the peak areas of the two analytes (BMA and PMA) and those for the deuterated internal standards (d_s-BMA and d_s-PMA) in the chromatograms. Divide the peak area of the analytes by the peak area from the matching deuterated internal standard.

23. Prepare calibration curves of the peak Area Std./Area Internal Std. (ratio calculated in step 22) versus the *urine equivalent* concentration of the standards for the two analytes.

CALCULATIONS:

24. Determine the concentration of the two analytes in the extracts from the original urine (4.0 mL specimen) from the curves obtained in step 23. The results can be expressed as ng/mL of each analyte in urine.

EVALUATION OF METHOD:

This method was evaluated and described in detail by B'Hymer [9,10]. A general summary of this published information is given below:

Accuracy and Precision. Three recovery studies using multiple columns over several days demonstrated the accuracy and precision of this test method. The first recovery study was performed using fortified urine samples over three separate experimental batch runs, and these data are presented in Table 2. Average recoveries were between 102 and 106% for the two analytes over the four spiked

concentration levels investigated. For each batch run, the experimental trial consisted of three samples at four different concentration levels. The recovery and precision for each level (n = 9 samples) are displayed in Table 2. The relative standard deviations (RSD) observed ranged from 2.0 to 5.0%. These accuracy and precision numbers fall well within parameters established for bioanalytical methods [7].

The second recovery study was performed using a urine substitute, fortified over three separate experimental batch runs, and these data are presented in Table 3. Average recoveries were between 99 and 109% for the two analytes over the four spiked concentration levels investigated. For each batch run, the experimental trial consisted of three samples at four different concentration levels. The recovery for each level (n = 9 samples) is displayed in Table 3. As seen in the RSD column of Table 3, precision was much worse for the urine substitute recovery study. This is due to a higher level of ion suppression in the electrospray source. This was especially pronounced at the lower spiked concentration levels (1 and 2 ng/mL). The highest result for the RSD of the 9 samples fortified at 1 ng/L BMA was 29%. While the accuracy numbers meet standard criteria, the precision values at the lower concentrations do not [7]. The use of a urine substitute for QC samples is not recommended for this method.

The third recovery study used urine specimens from twelve non-exposed volunteers; six were smokers and six were non-smokers. Most participants' specimens had base levels for both BMA and PMA as are shown in Table 4. Again, reasonable accuracy and precision were demonstrated for a 6 ng/mL equivalent level spike; individual recoveries ranged from 95 to 109% for BMA and 101 to 123% for PMA (Table 4). No interferences were detected in the unspiked urine from the 12 volunteer specimens. This study illustrates two important points: 1. Levels of both metabolites can be found in the urine of non-exposed individuals, which is why aqueous standards are used in this method, and 2. It is important to ascertain smoking status and other non-occupational exposures that may cause elevated levels of these metabolites.

Linearity. All calibration curves used during the development of this method were linear and had correlation coefficients of 0.99 and greater. The concentration range was equivalent to 0.5 to 50 ng/mL BMA and PMA for the extraction of 4.0 mL urine. Calibration curves were run at the beginning and end of all sample batch runs; calibration curve slope drift was found to be minimal.

Specificity. The optimized chromatographic conditions developed for this method, along with the tandem mass spectrometric detector, proved to be specific and showed no major interferences. The mass transition ions of BMA and PMA chosen in this method had the greatest response and were the predominant daughter ions.

Robustness. Two C18 HPLC columns (Zorbax Rx-C18, Agilent Technologies, Santa Clara, CA, USA) from the same manufacturer but from different manufacturing lots were used during the recovery studies. Accuracy and precision were not affected; therefore, the method was found to be reproducible with any normal functioning C18 HPLC column. Recovery results from individual volunteer urine specimens spiked with the analytes indicate that the method is accurate and not significantly affected by individual urine specimen matrix differences during analyte extraction. The urine substitute was found to cause problems with precision, especially at lower levels; this was attributed to an increase in ion suppression over human urine within the electrospray source.

Stability. Sample stability was evaluated, whereby a six-day stability study was conducted on the final chromatographic sample solution. BMA and PMA were stable at 8 °C (the autosampler temperature) and at room temperature in the absence of light. A light stability experiment was carried out by storing this solution in a clear glass vial at room temperature on the window sill, thus creating a worst-case scenario. After 1 day of storage in light under these conditions, BMA and PMA had mean assay values of 75 and 72% (n = 3), respectively, when compared to solutions of freshly prepared reference standards. After three days of light exposure, extensive degradation was noticed; BMA mean assay values had fallen to 9% (n = 3) of the original level and PMA had fallen to 16% (n = 3). After six days of exposure to light, both analytes were nearly completely degraded. Although both analytes benefit from the use of

individual deuterated internal standards, the use of amber glass autosampler vials or other means of reducing light exposure is recommended to ensure sample stability during extended chromatographic batch runs.

Range. This method should be considered accurate for the estimation of BMA and PMA in human urine within the 0.5 to 50 ng/mL standard curve range. Field samples at higher levels can be diluted to a concentration within that range for analysis.

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Analyte	Analyte added (ng/mL)	Mean concentration (ng/mL, n = 9)	Mean background corrected concentration (ng/mL, n = 9)	Average recovery (%)	Standard deviation (ng/mL)	%RSD ¹
BMA ²	1	7.43	7.18	103	0.37	5.0
BMA ²	2	8.44	8.18	103	0.19	2.3
BMA ²	8	14.6	14.2	103	0.32	2.2
BMA ²	30	38.2	36.2	106	0.94	2.4
PMA	1	1.02	1.00	102	0.05	4.9
PMA	2	2.10	2.00	105	0.09	4.3
PMA	8	8.23	8.00	103	0.27	3.3
PMA	30	31.9	30.0	106	0.65	2.0

Table 2. MULTIPLE LEVEL RECOVERY EXPERIMENT of BMA and PMA FROM URINE

Notes: 1. %RSD = percent relative standard deviation

2. The non-fortified reference urine had a background level of 6.2 ng/mL BMA and no detectable level of PMA.

Three different spiked samples were prepared at each level and analyzed during three separate experimental trial runs (a total of nine samples at each spike level were analyzed). The same C18 column was used for experimental batch trials 1 and 2; a second C18 column was used on trial run 3.

Analyte	Analyte added (ng/mL)	Mean concentration (ng/mL, n = 9)	Average recovery (%)	Standard deviation (ng/mL)	%RSD ¹
BMA	1	1.07	107	0.31	29
BMA	2	2.13	106	0.34	16
BMA	8	8.31	104	0.53	6.4
BMA	30	30.7	102	0.76	2.5
PMA	1	1.09	109	0.27	25
PMA	2	1.98	99	0.16	7.9
PMA	8	7.96	100	0.48	6.0
PMA	30	31.1	104	0.91	2.9

Table 3. MULTIPLE LEVEL RECOVERY EXPERIMENT of BMA and PMA FROM A URINESUBSTITUTE

Notes: 1. %RSD = percent relative standard deviation.

Three different spiked samples were prepared at each level and analyzed during three separate experimental trial runs (a total of nine samples at each spike level were analyzed).

	Backgrou	und Level	Fortif (backgrou	ied Urine nd + 6 ng/mL)	Fortified U	Jrine Recovery
Individual Sample	BMA (ng/ mL)	PMA (ng/ mL)	BMA (ng/ mL)	PMA (ng/mL)	BMA [ng/mL (percent)]	PMA [ng/mL (percent)]
Smoker 1	2.7	0.2	8.7	6.2	8.6 (99%)	6.5 (104%)
Smoker 2	28.3	0.3	34.3	6.3	34.7 (101%)	6.9 (108%)
Smoker 3	15.9	0.9	21.9	6.9	21.3 (97%)	7.9 (114%)
Smoker 4	5.7	0.3	11.7	6.3	11.6 (99%)	7.7 (121%)
Smoker 5	1.3	nd	7.3	6.0	6.9 (95%)	6.4 (106%)
Smoker 6	9.2	0.7	15.2	6.7	15.5 (102%)	7.0 (104%)
Mean	10.5	0.4	16.5	6.4	16.4 (99%)	7.1 (110%)
Non-smoker 1	0.3	nd	6.3	6.0	6.2 (98%)	6.2 (103%)
Non-smoker 2	7.1	nd	13.1	6.0	14.3 (109%)	7.4 (123%)
Non-smoker 3	6.8	nd	12.8	6.0	13.1 (102%)	6.8 (112%)
Non-smoker 4	23.3	nd	29.3	6.0	28.6 (97%)	6.3 (105%)
Non-smoker 5	4.7	nd	10.7	6.0	11.0 (103%)	6.3 (105%)
Non-smoker 6	7.2	nd	13.2	6.0	13.5 (102%)	6.1 (101%)
Mean	8.2	-	14.2	6.0	14.5 (102%)	6.5 (108%)

Table 4. INDIVIDUAL SMOKER and NON-SMOKER RECOVERY EXPERIMENT of BMA and PMA

Notes: The instrumental limit of detection (LOD) was estimated to be approximately 0.2 ng/mL for both analytes. The lowest standard value was 0.5 ng/mL for both analytes. [Values near the LOD were reported to one significant figure.] nd = none detected (< 0.2 ng/mL)